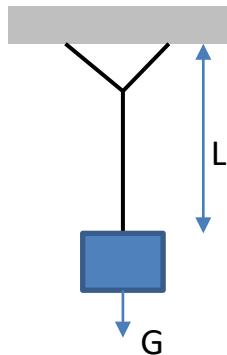


### Exercises 7: Viscoelasticity

**7.1:** A mass  $G$  is suspended to a plastic fiber. The fiber consists of an incompressible, viscoelastic material. In its unloaded state, the fiber has a cross section  $A_0$  and a length  $L_0$ .



The viscoelastic equation of state of the true fiber tension

$\sigma = G/A(\varepsilon)$  with  $A(\varepsilon)$  the instantaneous cross section  
and the extension  $\varepsilon = (L - L_0)/L_0$ ,  
 $L$  as instantaneous fiber length

is described by the Kelvin-Voigt Model with a spring constant  $E$  and a viscosity  $\eta$ .

For which mass  $G$  is the strain rate constant? What is then the resulting  $dL/dt$ ?

**7.2:** Use the Maxwell model analysis of stress relaxation to estimate the longest relaxation time for the highest molecular weight sample in the plot of the stress relaxation modulus  $G(t)$  versus time  $t$  on page 3 of the lecture notes [ $G = 3 \times 10^6$  dyn/cm<sup>2</sup> (plateau value)  $M = 4.6 \times 10^5$  g/mol,  $\eta = 5.5 \times 10^{11}$  P (= g/cm s)]

**7.3:** Use the BSM to estimate the longest relaxation time for polystyrene with  $M = 10^6$  g/mol in cyclohexane at the theta temperature (35°C). The values of  $D_t$  and  $[\eta]$  are given as a function of  $M$  for this system in Figure 1 and Figure 2 respectively. (viscosity of cyclohexane is 0.8 cP).

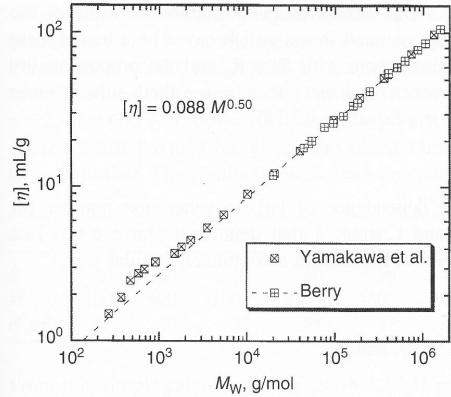


Figure 1: for  $M = 10^6$  use  $[\eta] = 84$  mL/g

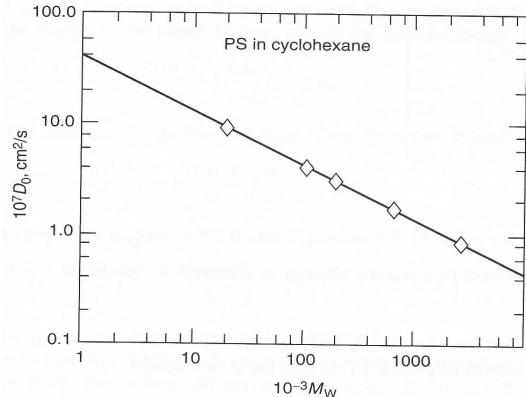


Figure 2: for  $M = 10^6$  is  $D_t \approx 10^{-7}$  cm<sup>2</sup>/s

**7.4:** You are an engineer in charge of a polystyrene drinking cup manufacturing plant. Normally, you process the material at 160°C, where the viscosity of the polymer melt is  $1.5 \times 10^3$  poises when  $Z_w = 800$  (Note that  $Z_{c,w}$  for polystyrene is 730). Today, your polystyrene has  $Z_w = 950$ . What change in processing temperature will bring down the viscosity to the initial value of  $1.5 \times 10^3$  poises?  $T_g$  for polystyrene is 100°C. (Note: Use the universal constants for the Williams-Landel-Ferry equation  $C_1 = 17.44$ ,  $C_2 = 51.6$ )

The following relationship exists between the melt viscosity of the polymer and its length:

$$\eta = K_H (Z_w)^{3.4}.$$

With  $Z$  the number of atoms along the polymer chain's backbone.

With  $Z_{c,w}$  as critical entanglement chain length: weight average number of chain atoms in the polymer molecules